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smallest representative unit of a substance.

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Chemistry > Chapter Seven > Chapter 7 Test Review. If you can do these, I guarantee you'll pass the test!!! Count atoms: $\text{Fe}(\text{OH})_2$ $\text{Na}_2(\text{SO}_4)_2$ $\text{Ca}_3(\text{PO}_4)_2$

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Chemistry Chapter 7 Test Multiple Choice Identify the choice that best completes the statement or answers the question. ____ 1. A chemical formula includes the symbols of the elements in the compound and subscripts that indicate a. atomic mass of each element. b. number of atoms or ions of each element that are combined in the compound. c ...

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

Holt Modern Chemistry Chapter 7 Section 1 Review Answers

This video explains the answers to the practice quiz on Chapter 7 - 8, which can be found here: <https://goo.gl/yxsQH4>.

Chapter 7 - 8 Practice Quiz

CHEM Chapter 7- Classifying Chemical Reactions. Fall 2015. Roman Czernuszewicz. CHEM 1301. These notes cover Chapter 7 from "Introductory Chemistry" by Zumdahl and Decoste, 7th/8th edition. They include all important vocabulary, examples, and

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side notes.

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