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Chemistry: Mixtures and Solutions.
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that are homogeneous mixtures of 2 or more substances in a single phase.

Solute. the part of a solution that gets dissolved that could be any state of matter. Solvent. the part of a solution that does the solving.

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SOLUTIONS are homogeneous mixtures.
A solution is a mixture of two or more

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substances in a single phase. At least two substances must be mixed in order to have a solution. The substance in the smallest amount and the one that dissolves or disperses is called the SOLUTE.

**What are Mixtures and Solutions? -
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This activity will teach students about the characteristics of mixtures and the differences between homogeneous and

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heterogeneous mixtures.

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Chemistry II: Chapter 14 Mixtures and Solutions. a type of heterogeneous mixture whose particles settle out over time and can be separated from the mixture by filtration. a heterogeneous

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mixture of intermediate-sized particles (between atomic-size of solution particles and the size of suspension particles).

Chemistry II: Chapter 14 Mixtures and Solutions ...

In chemistry a solution is actually a type of mixture. A solution is a mixture that is

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the same or uniform throughout. Think of the example of salt water. This is also called a "homogenous mixture."

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In chemistry, a solution is a homogenous mixture (equal throughout) containing two or more components. A substance, or solute, is dissolved into another

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substance, known as the solvent.

What is a solution in chemistry? | Study.com

Mixtures and solutions are part of everyday life which makes the topic relatable and obtainable to keep student engagement. The activities pertaining to mixtures and solutions will not be solely

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done during science class, but instead the unit will also be incorporated into social studies, art, math, vocabulary, and music.

Unit Plan: Mixtures and Solutions Fifth Grade

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Chemical solutions are usually

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homogeneous mixtures. The exception would be solutions that contain another phase of matter. For example, you can make a homogeneous solution of sugar and water, but if there are crystals in the solution, it becomes a heterogeneous mixture. Many common chemicals are homogeneous mixtures.

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